

Candidate Name _____

Centre Number		Candidate Number									

EXAMINATIONS COUNCIL OF ZAMBIA

Joint Examination for the School Certificate
and General Certificate of Education Ordinary Level

SCIENCE

5124/2

PAPER 2 (PHYSICS)

Wednesday

2 NOVEMBER 2011

1 hour 15 minutes

Additional materials:
Mathematical tables
Graph paper
Writing paper

Time: 1 hour 15 minutes

INSTRUCTIONS TO CANDIDATES

Write your name, centre number and candidate number at the top of this page and on any separate answer paper used.

There are **twelve (12)** questions in this paper.

Section A

Answer **all** the questions.

Write your answers in the spaces provided on the question paper.

Section B

Answer any **two** questions.

Write your answers on the Answer Booklet provided.

At the end of the examination

1. Fasten Answer Booklet used securely to the question paper.
2. Enter the numbers of the **Section B** questions you have answered in the grid below.

INFORMATION FOR CANDIDATES

The number of marks is given in brackets [] at the end of each question or part question.

Cell phones are not allowed in the Examination room.

Candidate's use	Examiner's use
Section A	
Section B	
Total	

Section A

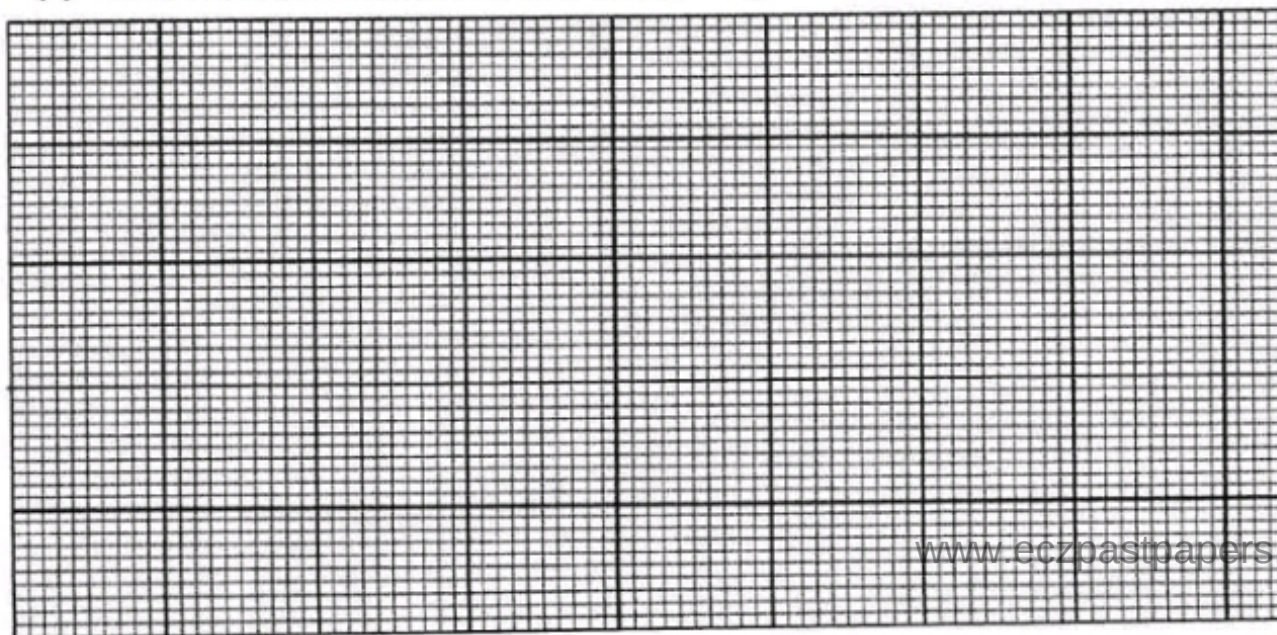
[45 marks]

Answer all questions.

Write your answers in the spaces provided on the question paper.

- 1 A cyclist accelerates uniformly from rest to a velocity of 10m/s in 3s . He then moves with a constant velocity of 10m/s for 6s , before decelerating uniformly to rest in a further 5s .

(a) Draw a velocity-time graph representing the cyclist's motion.



[3]

(b) Calculate the distance travelled by the cyclist during the journey.

Distance = [2]

Total: [5]

- 2 (a) What is meant by the centre of mass of an object?

[1]

- (b) Explain why a minibus is more likely to topple over when the roof rack is heavily loaded than when the roof rack is empty.

[2]

- (c) A metre rule is supported on a knife-edge placed at the 40cm mark. It is found that the metre rule balances when a mass of 45g is suspended from the 15cm mark as shown in **Figure 2.1**.

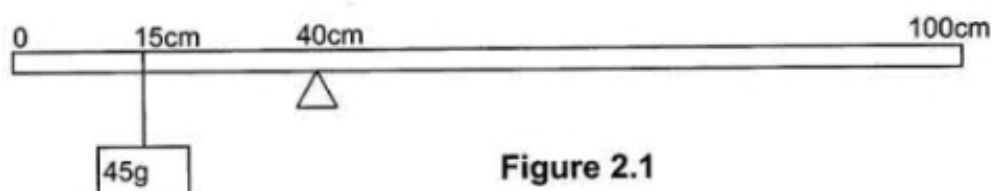


Figure 2.1

If the centre of mass of the metre rule is at the 52.5cm mark, calculate the mass of the metre rule.

Mass = [2]

Total: [5]

- 3 A girl whose mass is 45kg carries a box of mass 25kg up a flight of steps. There are 40 steps each 40cm high. She takes 16s to climb up the steps. (Take $g = 10\text{m/s}^2$)

- (a) What is her weight?

Weight = [1]

- (b) What is the weight of the box?

Weight = [1]

- (c) Calculate the total gravitational potential energy of the girl and the box when she reaches the top.

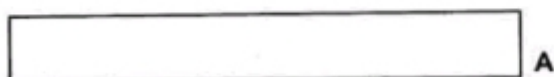
Gravitational potential energy = [2]

- (d) Calculate the total power.

Power = [2]

Total: [6]

- 4 (a) The diagram below represents a bar of soft iron which is to be magnetised with a north pole at end **A**.



Complete the diagram to show how you would arrange a coil connected to a cell to achieve this. Make clear on your diagram the windings of the coil and the polarity of each cell terminal. [3]

- (b) What would be the effect of disconnecting the cell on the magnetisation of the bar?

_____ [1]

Total: [4]

- 5 (a) The pressure of air in a tyre of an empty lorry is $3.0 \times 10^5 \text{ Pa}$ and the volume of the air in the tyre is 0.080 m^3 . Calculate the volume of the air in the tyre when the lorry is loaded until the pressure of the air in the tyre rises to $3.6 \times 10^5 \text{ Pa}$. (Assume that the air temperature is constant).

Volume = [3]

- (b) The tyre pressure of a lorry that has been moving for sometime is usually greater than the pressure when the lorry has been standing at rest. Why is this so?

[2]

Total: [5]

- Two successive crests of an approaching water wave are separated by a distance of 1.5m. It takes 0.2s for one crest to cover the distance of 1.5m.

- (a) At what speed is the wave travelling?

Speed =

[2]

- (b) What is the frequency of the wave?

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Frequency =

[2]

- (c) Distinguish between the nature of water wave and a sound wave.

(i) Water wave is _____

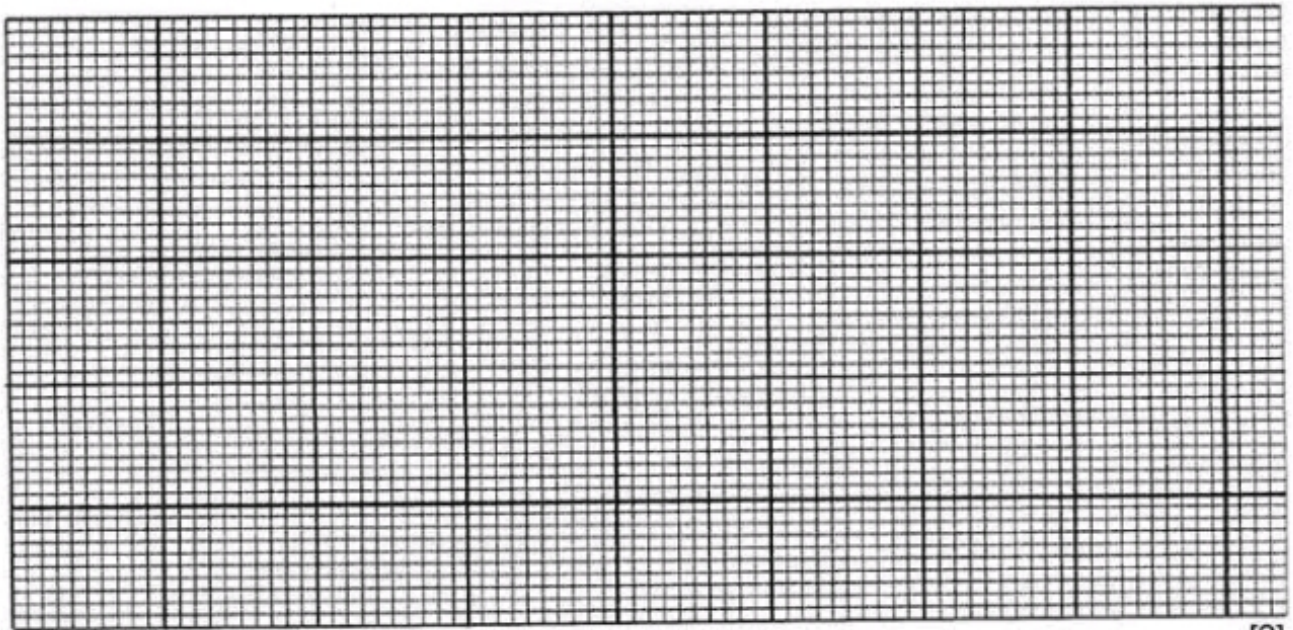
[1]

(ii) Sound wave is _____

[1]

Total: [6]

- 7 A 4cm high object is placed 8cm from a convex lens of focal length 3cm. Draw a scale diagram to find the position and size of the image.



[2]

- (a) Position of the image

Position =

[1]

- (b) Size of the image

Size =

[1]

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Total: [4]

- 8 Figure 8.1 shows a diagram of a clinical thermometer with some features labelled.

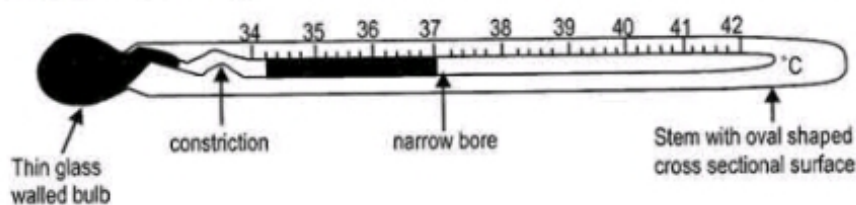


Figure 8.1

Explain why it has each of the following features:-

- (a) A thin glass walled bulb,

[1]

(b) A constriction,

 _____ [1]

(c) A short range of temperature calibration,

 _____ [1]

(d) A narrow bore,

 _____ [1]

(e) An oval shaped glass stem.

 _____ [1]

Total: [5]

- 9 **Figure 9.1** shows two metal spheres **A** and **B** being charged by induction using a negatively charged rod brought near **A**. In **figure 9.2**, the two spheres are separated. In **figure 9.3**, the charged rod is removed.

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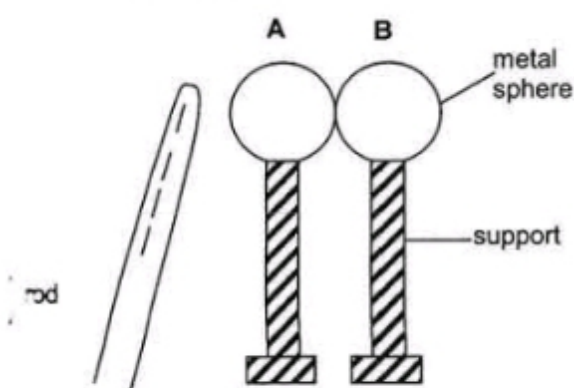


Figure 9.1

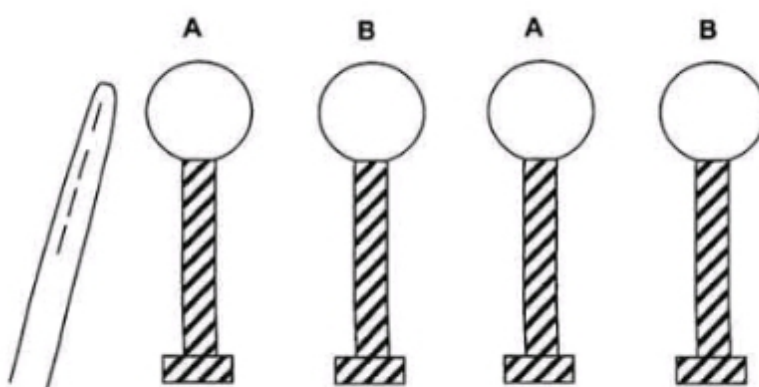


Figure 9.2



Figure 9.3

(a) What should be the nature of material used as the support of each sphere?

_____ [1]

(b) On each sphere in 9.1, 9.2 and 9.3, indicate the charge distribution.

[4]

Total: [5]

Section B

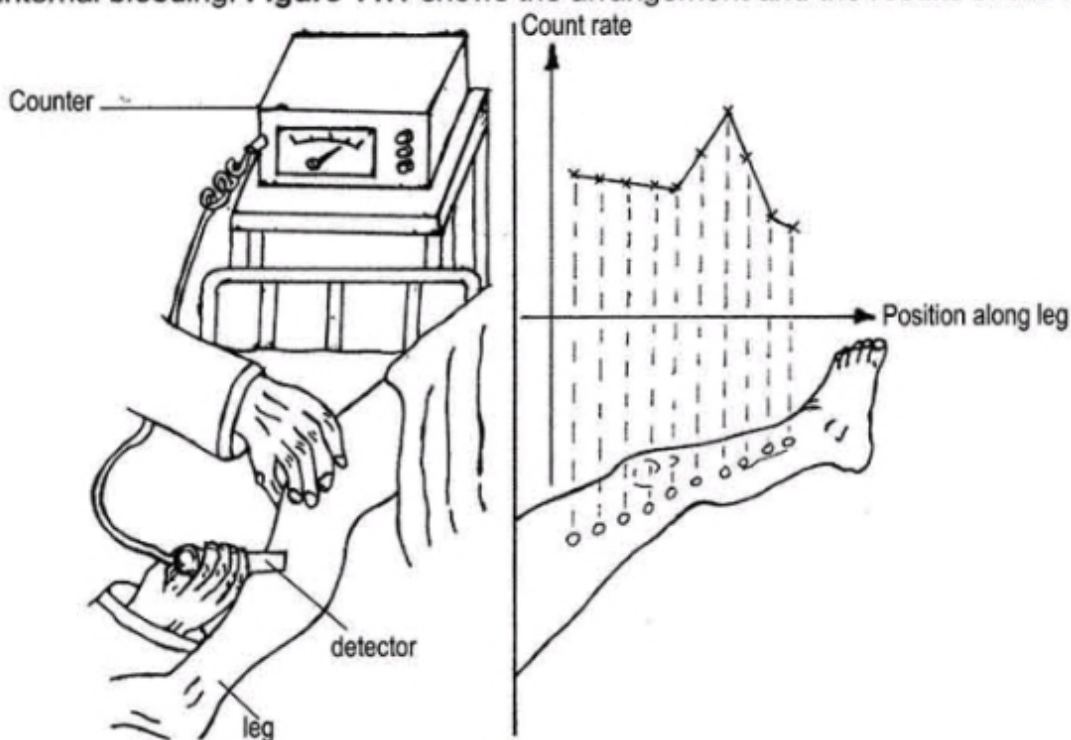
[20 marks]

Answer any **two (2)** questions from this section.Use the **Answer Booklet** provided.

- 10 (a) Define density of a substance. [1]
- (b) Describe an experiment you would perform to determine the density of an irregularly shaped object such as a stone. [5]
- (c) An empty tin of mass 20g and capacity 25cm^3 was used to measure the density of mercury. When full of mercury, the mass of the tin and mercury was 360g. What is the density of mercury? [4]

Total: [10]

- 11 Radioactive isotopes can be used to locate internal bleeding in the body. A commonly used radioactive isotope is iodine – 131 (^{131}I). This emits gamma radiation and has a half-life of 8 days.
- (a) Describe what is meant by a radioactive isotope. [2]
- (b) Describe what is meant by half-life. [2]
- (c) The activity of a sample of iodine – 131 (^{131}I) was measured over a period of 20 minutes on three separate occasions. The readings obtained were:- 338 β q, 326 β q, 356 β q.
- (i) Explain why the readings were not all the same. [1]
- (ii) Calculate the average count rate for the radioactive isotope. [1]
- (d) A patient has internal bleeding from a blood vessel in her leg. A small quantity of the isotope ^{131}I is injected into her blood stream. A detector is used to find the internal bleeding. **Figure 11.1** shows the arrangement and the results of the test.

**Figure 11.1**

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- (i) State the name of a suitable detector. [1]
- (ii) The radioactive isotope used for this purpose is a gamma emitter and not an alpha emitter. Why is a gamma emitter used? [2]
- (iii) How will the doctor tell from the results where the internal bleeding is taking place? [1]

Total: [10]

- 12 **Figure 12.1** shows a bucket with a heating element at the bottom. The bucket has a plastic body with the outside casing of the element which is a metal.

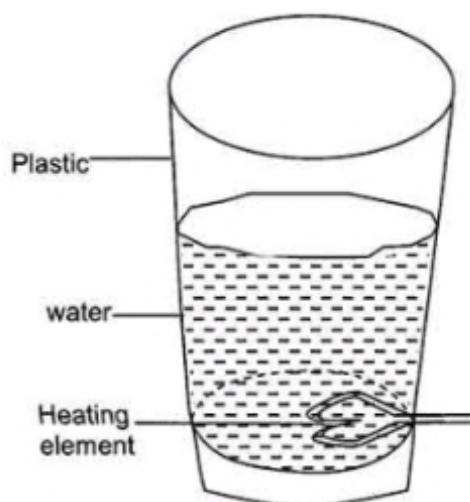


Figure 12.1

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- (a) Explain why;
 - (i) a fuse is included in the circuit and what happens when the fuse blows. [2]
 - (ii) the metal casing of the heating element is connected to the earth. [2]
- (b) The electrical power input to the bucket is 2000W and the bucket is switched on for 6 minutes. Calculate the electrical energy, in Joules supplied to the bucket. [3]
- (c) Explain, in molecular terms, how evaporation causes loss of energy from the water when the element is switched off. [3]

Total: [10]